

I Knowledge of course

1. Ground state and 2 first singlet + 2 first triplet excited states of He:

$1s^2$ :  $^1S_0$  ground

$1s2s$ :  $^1S_0$  singlet

$3S_1$  triplet (lower in energy)

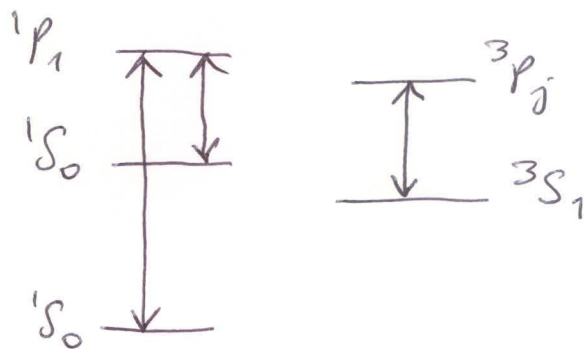
$1s2p$ :  $^1P_1$  singlet

$3P_j$  triplet  $j = 0, 1, 2$

allowed if:

$\Delta S = 0$

$\Delta L = \pm 1$



2. Be ( $Z=4$ )  $1s^2 2s^2$   $^1S_0$

B ( $Z=5$ )  $1s^2 2s^2 2p^1$   $^2P_{1/2}$

C ( $Z=6$ )  $1s^2 2s^2 2p^2$   $^3P_0$

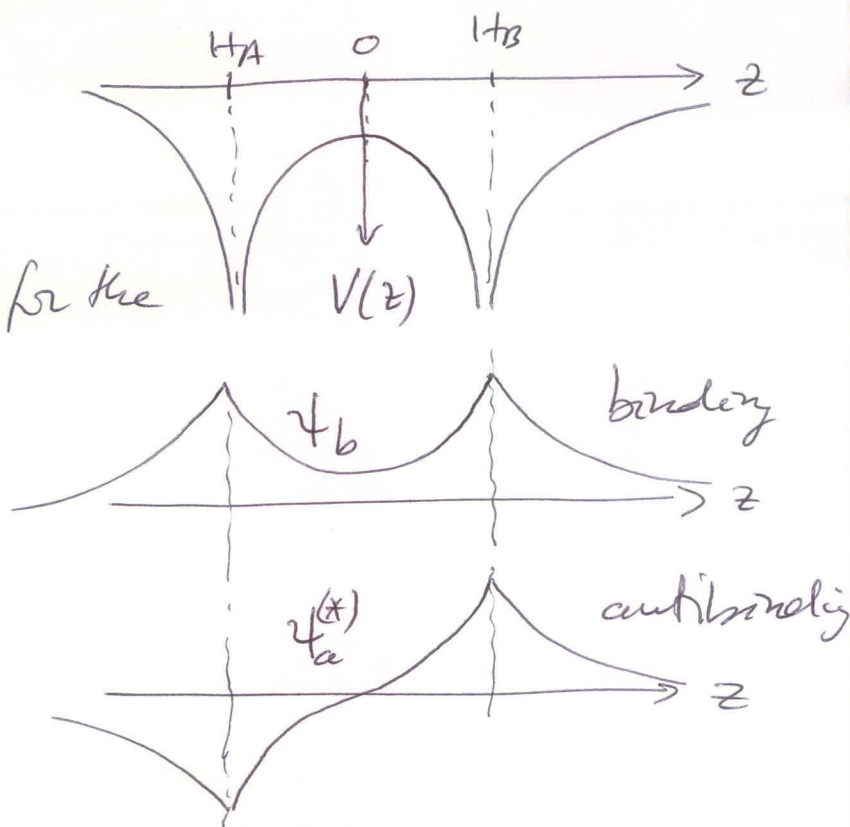
less than half-filled: lowest  $j$  has lower energy.

3. Al  $Z = 13$

Ground state  $1s^2 2s^2 2p^6 3s^2 3p^1$

4.  $H_2^+$  mol. ion

wavefunction of  $e^-$  on internuclear axis for the 2 lowest states:



$$\psi_b(z) \approx \psi_A(z) + \psi_B(z)$$

$$\psi_a^{(*)}(z) \approx \psi_A(z) - \psi_B(z)$$

$\psi_b(z)$  is the wavefunction with the lowest energy, as deduced from perturbation theory of  $V_B$  for H-atom around  $H_A$  (and vice-versa).

5. energy spacing

electronic  $>$  vibrational  $>$  rotational

(the ratio is about  $\sqrt{\frac{m_e}{M}}$ ).

6. Rotational transitions appear between levels

$BK(K+1)$  with  $\Delta K = \pm 1$ . Therefore, in absorption  $B(K+1)(K+2) - B(K+1)K = 2B(K+1)$  with  $K=0,1,2, \dots$ . Lines, equally spaced by  $2B$



Here  $2B \approx 20 \text{ cm}^{-1} \Rightarrow B = 10 \text{ cm}^{-1}$

and  $B = \frac{h^2}{2J}$   $J$  moment of inertia.

$J = \frac{h^2}{2B}$

7. Shells 1, 2, 3 are full - shell four has the config.  $4s^2 4p 4d$  -

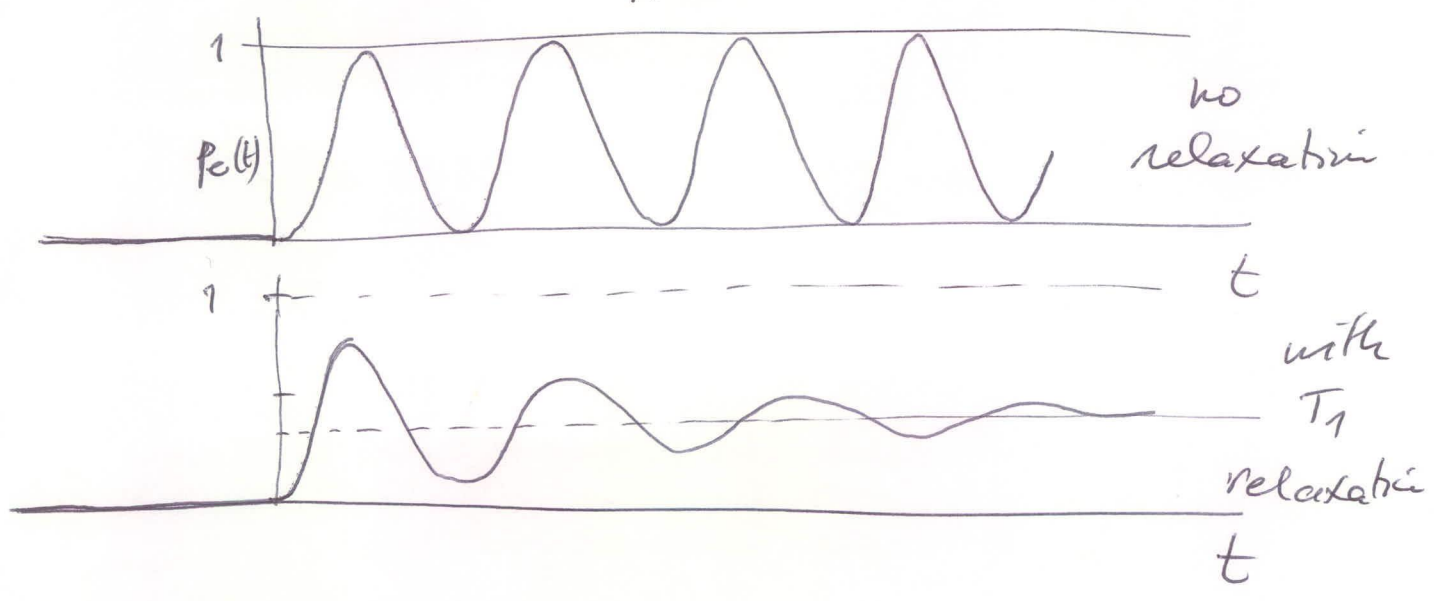
$4s^2$  is a closed sub-shell, thus we consider only the 2 electrons  $4p 4d$ . Pauli principle doesn't apply, so all combinations are allowed:

$2S+1 = 1$  (singlets)  $L = 2-1, 2, 2+1$  (1, 2, 3)  
 $2S+1 = 3$  (triplets)  $L = 1, 2, 3$

$^1P, ^1D, ^1F, ^3P, ^3D, ^3F$  (not ground state).

The ground state has  $4s^2 4p^2$  (term  $^3P_0$ ).

8. Optical mutation: apply  $\vec{E}_0 \cos \omega t$  at  $t=0$



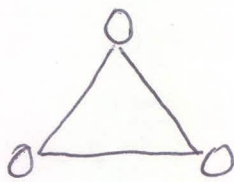
9. Population difference  $p_g - p_e$  with

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$$\frac{p_e}{p_g} = \exp\left(-\frac{h\nu}{k_B T}\right) \quad \text{here } |p_e - p_g| \ll 1 \text{ thus}$$

$$(p_g - p_e) \approx p_g \left(1 - \exp\left(-\frac{h\nu}{k_B T}\right)\right) \approx \frac{1}{2} \cdot \frac{h\nu}{k_B T}$$

$$\frac{1}{2} \frac{h\nu}{k_B T} = \frac{1}{2} \frac{6.6 \cdot 10^{-34} \times 43 \cdot 10^6}{1.38 \cdot 10^{-23} \times 300} = \frac{1}{2} \frac{43 \times 6.6}{4.4 \cdot 10^{-21}} \cdot 10^{-28} \approx 3.3 \cdot 10^{-6}$$

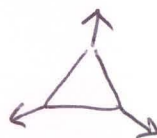
10.  number of modes  $3N - 6$  (not linear) = 3 modes of vibration.

All modes have to be in-plane motions

because there are  $3 \times 2 - 2 - 1 = 3$  in-plane vibrations.

2 transl. 1 rotation

• symmetric mode



• 2 degenerate antisymmetric modes



[ note: these 2 modes sketched are not orthogonal ]



## II Problem: Hyperfine structure

(5)

1.  $I$  nuclear angular mom. number

$I^2$  has eigenvalues  $I(I+1) \hbar^2$

$I_z$  —————  $m_I \hbar$ ,  $m_I = -I, -I+1, \dots, I-1, I$ .

Spin degeneracy  $2I+1$ .

$^{133}\text{Cs}$  has  $I = \frac{7}{2}$   $I(I+1) = \frac{63}{4}$

$m_I = -\frac{7}{2}, -\frac{5}{2}, -\frac{3}{2}, -\frac{1}{2}, \frac{1}{2}, \frac{3}{2}, \frac{5}{2}, \frac{7}{2}$

$2I+1 = 8$ .

2.  $\vec{j} = \vec{l} + \vec{s}$ ,  $j$  total electron ang. mom.

$\vec{F} = \vec{j} + \vec{I}$   $F$  total atomic ang. mom.

(a) possible values of  $F = |I-j|, |I-j|+1, \dots, I+j$ .

degeneracy is  $2F+1$  for each value of  $F$ .

(b) if  $I+j$  is integer, then total degeneracy is

a sum of odd integers:

$$2(I+j)+1 + 2[(I+j)-1]+1 + \dots + 2[|I-j|]+1 = N$$

$$\text{but } 2p+1 + 2(p-1)+1 + \dots + 1 = (p+1)^2$$

$$\text{Thus } N = (I+j+1)^2 - (|I-j|)^2 = (I+j+1-I+j)(I+j+1+I-j)$$

$$= (2j+1)(2I+1)$$

Let us consider now the case  $I+j = \text{half-integer}$   
and assume  $I \geq j$  (without loss of generality)

Set  $I+j = p+\frac{1}{2}$

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The sum of all degeneracies from  $\frac{1}{2}$  to  $I+j$  is:

$$2+4+\dots+2(I+j)+1 = 2(1+2+\dots+p+1) = (p+1)(p+2)$$

From  $\frac{1}{2}$  to  $|I-j|$ , this sum is  $(q+1)(q+2)$

if  $|I-j| = q+\frac{1}{2}$

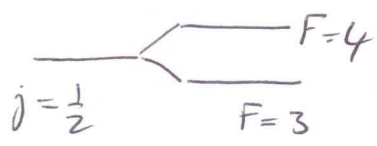
Thus  $N = \text{sum}(I+j) - \text{sum}(|I-j|-1) = (p+1)(p+2) - q(q+1)$

$$= \underbrace{(I+j+\frac{1}{2})(I+j+\frac{3}{2})}_{(I+j)^2+2(I+j)+\frac{3}{4}} - \underbrace{(I-j-\frac{1}{2})(I-j+\frac{1}{2})}_{(I-j)^2-\frac{1}{4}} = (I+j)^2+2(I+j)+1 - (I-j)^2$$

$$N = 2I \times 2j + 2I + 2j + 1 = (2I+1)(2j+1)$$

(c)  $^{133}\text{Cs}$  has the config. of a rare gas (Xe) plus one  $e^-$  (it is an alkali). We may thus consider only that electron -

$I = \frac{7}{2}, j = \frac{1}{2} \rightarrow F = 4 \text{ or } 3$

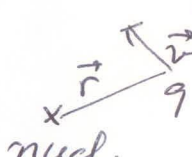


3. Biot-Savart law:  $\vec{B} = \frac{\mu_0}{4\pi} \frac{q\vec{v} \times \vec{R}}{R^3}$

Field created by the orbital ang. mov. of the electron at the nucleus' position:  $\vec{R} = -\vec{r}$

$\vec{B}_0 = -\frac{\mu_0}{4\pi} \frac{q\vec{v} \times \vec{r}}{r^3} = +\frac{\mu_0}{4\pi} \frac{q}{m} \frac{\vec{l}}{r^3}$  with  $\vec{l} = m\vec{r} \times \vec{v}$

$H = -\vec{\mu}_l \cdot \vec{B}_0 = -\frac{\mu_0}{4\pi} \frac{q}{m r^3} \vec{l} \cdot \vec{\mu}_I$



4. Spin magnetic moment

$$\vec{\mu}_S = -g_S \vec{S} \cdot \mu_B$$

$$\vec{B}_{el} = \vec{B}_0 + \vec{B}_S$$

$$\vec{B}_S = \frac{\mu_0}{4\pi r^3} \left[ -\vec{\mu}_S + 3 \frac{\vec{\mu}_S \cdot \vec{r}}{r^2} \vec{r} \right]$$

$$H_{int} = -\vec{\mu}_I \cdot \vec{B}_{el} = -\vec{\mu}_I \cdot \left[ + \frac{\mu_0}{4\pi} \frac{q}{m} \frac{\vec{l}}{r^3} + \frac{\mu_0}{4\pi r^3} \left( -\vec{\mu}_S + 3 \frac{\vec{\mu}_S \cdot \vec{r}}{r^2} \vec{r} \right) \right]$$

$$= \vec{\mu}_I \cdot \left[ -\frac{q}{m} \frac{\vec{l}}{r^3} + \frac{1}{r^3} \left( \vec{\mu}_S - 3 \frac{\vec{\mu}_S \cdot \vec{r}}{r^2} \vec{r} \right) \right] \frac{\mu_0}{4\pi}$$

$$= \frac{\mu_0}{4\pi r^3} \cdot \frac{q}{m} \vec{\mu}_I \cdot \left[ -\vec{l} + \frac{g_S}{2} \left( \vec{S} - 3 \frac{\vec{S} \cdot \vec{r}}{r^2} \vec{r} \right) \right]$$

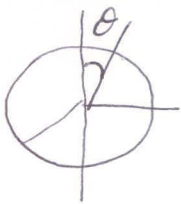
5. Averaging the Hamiltonian close to 0 in perturbation theory, we get

$$\int 4\pi r^2 dr |\psi|^2 H(r)$$

$$H(r) \sim r^{-3}, |\psi|^2 \sim r^{2l} \Rightarrow \int r^{2-3+2l} dr = \int r^{2l-1} dr$$

converges for  $l \geq 1$ , but diverges for  $l=0$ !

(a) Average of the dipole field outside the nucleus



angular integration:

$$\int d\varphi \int \sin\theta d\theta \cdot \underbrace{(1 - 3\cos^2\theta)}_{H_{int}} = 2\pi \int d\varphi (1 - 3\cos^2\theta) = 0$$

(b) normal component of  $B$  continuous follows

from the eq.  $\vec{\nabla} \cdot \vec{B} = 0$  ( $\text{div} \cdot \vec{B} = 0$ )



$$\vec{B}_i = \vec{B}_{dipole} \text{ on axis at } r=a = \frac{+2\mu_I \mu_0}{4\pi a^3}$$

(c) The average interaction is 0 outside the sphere and constant inside because the magnetic field is uniform. Thus:

$$H_{int} = -\vec{\mu}_{el} \cdot \vec{B}_I = -\vec{\mu}_{el} \cdot \frac{2\mu_0 \vec{\mu}_I}{4\pi a^3} \times \frac{4\pi}{3} a^3 \delta(\vec{r})$$
$$= -\frac{1}{3} 2\mu_0 \vec{\mu}_{el} \cdot \vec{\mu}_I \delta = -\frac{2}{3} \mu_0 \vec{\mu}_s \cdot \vec{\mu}_I \delta(\vec{r})$$

because only  $\vec{\mu}_s$  contributes to the magnetic moment of the electron ( $\vec{l}=0$ ).

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